

Hydrozirconation of Allenyl Sulfide for Generating γ -Thio-substituted Allylzirconium Species and Its Reaction with Carbonyl Compounds

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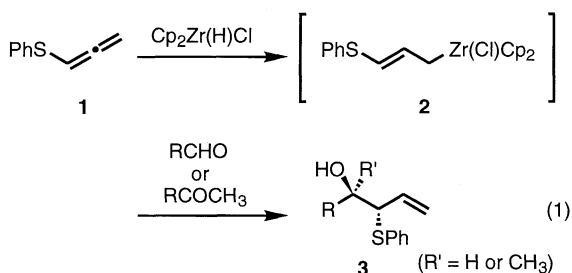
Hydrozirconation of allenyl sulfide generates a γ -thio-allylzirconocene species, which reacts with aldehydes and ketones to give anti olefinic β -sulfenyl alcohols in highly regio- and stereoselective manner.

γ -Thio-substituted allylmetals **A** are among the versatile classes of organosulfur reagents in organic synthesis.^{1,2} Based on our previous finding on the accessibility to various *allylzirconocene species via the hydrozirconation of allenes*,³ we envisioned to generate the Zr-version of **A** [$M = \text{Cp}_2\text{Zr}(\text{Cl})$] by applying the above protocol to thio allene **B**.

Figure 1.



In this communication, we wish to report the viability of such an approach to γ -thioallylzirconocene chloride **2** and the reaction with carbonyl compounds to give olefinic β -sulfenyl alcohols **3** in high stereoselectivity (Eq. 1).



Allenyl sulfide **1**⁴ underwent clean hydrozirconation by a similar procedure we previously reported,³ and the resulting thioallylzirconium **2** smoothly reacted with aldehydes at -78°C . The addition proceeded with high stereoselectivity to give the *anti*-adducts in high yield, as had been the case for various other allylzirconiums (Table 1).^{3,5,6} The addition to α,β -unsaturated aldehyde proceeded in a 1,2-mode to give **3d**, and none of the 1,4-adduct was observed (run 4). The olefinic β -sulfenyl alcohols, thus obtained in high stereoselectivity, were easily converted to vinyl epoxides, which served to confirm the stereostructure of the major adducts.⁶ Furthermore, at the temperature of 0°C , the thioallylzirconium **2** proved to react with ketones to give high yields of the adducts **3e–3g**.^{5,7} In case the sterics of the two substituents attached to the ketone carbonyl were sufficiently different, good to high anti selectivity was observed as seen in runs 6 and 7.⁷

Typical procedure is described for the synthesis of β -sulfenyl alcohol **3b**: Under a nitrogen atmosphere at -78°C , to a stirred suspension of $\text{Cp}_2\text{Zr}(\text{H})\text{Cl}$ ⁸ (322 mg, 1.25 mmol) in

Table 1. Reaction of **2** with aldehydes and ketones^a

run	aldehyde or methyl ketone	product ^b (yield/%, anti/syn)
1	$n\text{-C}_5\text{H}_{11}\text{CHO}$	 3a (75, 93/7)
2		 3b (81, >97/3)
3		 3c (80, 95/5)
4	$n\text{-C}_3\text{H}_7\text{CH=CHCHO}$	 3d (80, 86/14)
5	$n\text{-C}_{13}\text{H}_{27}\text{COCH}_3$	 3e (81, 82/18)
6		 3f (80, 96/4)
7		 3g (94, 88/12)

^aReaction conditions: -78°C , 30 min for runs 1–4; 0°C , 1 h for runs 5–7. ^bEach of the major stereoisomers is depicted.

CH_2Cl_2 (1.0 ml) [for a special precaution to prepare the suspension, see Ref. 3b] was added allenyl sulfide **1**⁴ (330 mg,

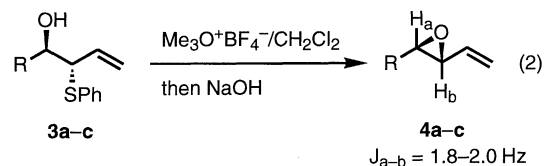
2.23 mmol) in CH_2Cl_2 (2.5 ml). The reaction was gradually warmed up to room temperature, where a red solution resulted. The reaction was rechilled to -78°C , to which a solution of cyclohexanecarbaldehyde (98.7 mg, 0.881 mmol) in CH_2Cl_2 (2.5 ml) was added. After 30 min, the reaction was stopped by adding sat. aqueous NaHCO_3 . Extractive workup followed by purification (SiO_2 preparative TLC, hexane/ Et_2O = 9/1) gave alcohol **3b** (186 mg, 81%).^{5,6}

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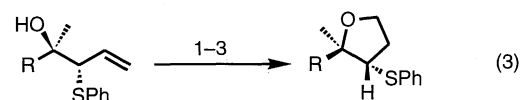
References and Notes

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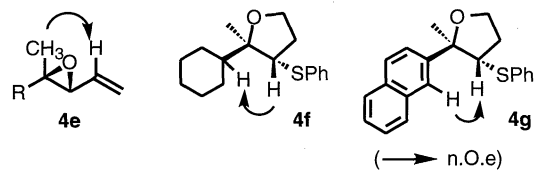
- 5 All new compounds were fully characterized by ^1H , ^{13}C NMR, IR, and high-resolution MS.
- 6 The stereoselectivities were determined by the peak integration of ^1H NMR (400 MHz). The stereochemistry was assigned by converting the major adducts **3a-c** to the corresponding vinyl epoxides **4a-c** as shown below.



- 7 The stereoselectivities were determined by the peak integration of ^1H NMR (400 MHz). The stereochemistry was assigned by the NOE experiments for vinyl epoxide **4e**, obtained from **3e** in a similar manner as in Eq. 2, and for tetrahydrofurans **4f** and **4g**, which were obtained from **3f** and **3g** (Eq. 3), respectively.



(1) 9-BBN; (2) H_2O_2 , NaOH; (3) Tf_2O , 2,6-di-*tert*-butyl-4-methylpyridine.



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